Two New Cytotoxic Tetracyclic Tetraterpenoids from the Soft Coral Sarcophyton tortuosum

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Received January 30, 2004

Two new cytotoxic tetracyclic tetraterpenoids, methyl tortuoate A (1) and methyl tortuoate B (2), along with the known methyl sartortuoate (3) were isolated from the soft coral *Sarcophyton tortuosum*. The structures of 1 and 2 were established by spectroscopic methods, mainly on the basis of 2D NMR techniques, and were confirmed by single-crystal X-ray diffraction analysis. The cytotoxic activities of these compounds were carried out in vitro on human nasophyringeal carcinoma (CNE-2) and murine lymphocytic leukemia (P-388) tumor cell lines.

In the course of our study on the chemical constituents of soft corals, we have isolated many chemically and biologically interesting secondary metabolites.¹⁻⁴ Among them, two tetraterpenoids, named methyl sartortuoate $(3)^5$ and methyl isosartortuoate (4),⁶ from the soft coral Sarcophyton tortuosum were structurally the most interesting compounds. After inspection of the structures of 3 and 4, we proposed that this unique framework might be derived from Diels-Alder addition of two cembranoid units as precursors. Later, two other biscembranoids, methyl sarcophytoate and methyl chlorosarcophytoate, together with the diterpenoid methyl sarcoate, one of the proposed biosynthetic precursors, from S. glaucum were reported.^{7,8} The structure of methyl sarcoate was remarkably similar to the biosynthetic precursor that we had proposed previously. In addition, Bowden et al. isolated a biscembranoid, methyl neosartortuate acetate, and methyl sarcoate along with another precursor from S. tortuosum.⁹ These findings supported the biogenetic pathway of this type of biscembranoid.

Recently,we isolated two more new biscembranoids, name methyl tortuoate A (1) and methyl tortuoate B (2), along with the known methyl sartortuoate (3), from the soft coral *S. tortuosum* Tix.-Dur. (Alcyoniidae) collected from Sanya Bay Hainan Island of China, in 2002. Herein, we report the isolation and structural determination of these two new biscembranoids.

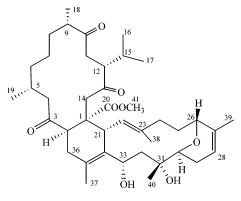
Compound 1 was obtained as colorless crystals, mp 242– 244 °C, $[\alpha]^{25}_{\rm D}$ +348° (*c* 0.025, CHCl₃). HRFABMS [*m/z* 705.4577 (M + Na)⁺] revealed a molecular formula of C₄₁H₆₂O₈. The IR spectrum (KBr) indicated the presence of absorptions characteristic of hydroxyl groups (3381, 1104 cm⁻¹), an ester carbonyl group (1742, 1194 cm⁻¹), and ketone carbonyl groups (1707 cm⁻¹). Its NMR spectra (Table 1) showed the following functionalities: one methyl ester [$\delta_{\rm C}$ 174.8, C-20; 51.4, C-41; $\delta_{\rm H}$ 3.60 (3H, s, H-41)]; three ketone carbonyl groups ($\delta_{\rm C}$ 213.6, C-10; 212.6, C-13; 212.0 C-3), two tri- and one tetrasubstituted double bond $[\delta_{\rm C}$ 141.0, C-23; 134.4, C-27; 132.8, C-34; 129.1, C-35; 122.9, C-22; 120.2, C-28; $\delta_{\rm H}$ 5.12 (1H, d, J = 10.0 Hz, H-22) and 5.55 (1H, dd, J = 4.5, 1.5 Hz, H-28)]; three allylic methyl groups $[\delta_{\rm H}$ 1.64 (3H, s, H-39), 1.86 (3H, s, H-38), and 1.91 (3H, s, H-37)]; and an isopropyl group $[\delta_{\rm H}$ 0.80 (3H, d, J = 7.0 Hz, H-15), 0.94 (3H, d, J = 7.0 Hz, H-17)]. In addition, four carbons were attached to oxygen ($\delta_{\rm C}$ 79.7, C-26; 70.1, C-33; 68.3, C-30; 74.2, C-31), but only three oxygen atoms remained, plausibly as one ether linkage and one tertiary and one secondary hydroxyl. The presence of nine methyl signals and spectroscopic features that included the ketone and methyl ester groups led us to recognize a tetracyclic biscembranoid similar to those we had reported previously.^{5,6}

From detailed analysis of the ¹H and ¹³C NMR data associated with C-H one-bond interactions and from crosspeaks observed in the ¹H-¹³C 2D NMR shift-correlated spectrum, all the signals of H and C could be assigned (Table 1). The correlations of ¹H-¹H COSY revealed seven proton-proton networks, as shown in Figure 1. The carbon connectivity showed that compound 1 had the same carbon skeleton as that of methyl sartortuate (3). The structure of 1 was similar to that of 3, except for the absence of the conjugated diene and one tertiary hydroxyl group. Instead, a new secondary hydroxyl was present. The HMBC connectivity, H-33/C-34, C-21, and C-31, allowed a decision for the placement of this secondary hydroxyl. Its HMBC spectrum also showed many informative ¹H-¹³C long-range correlations, such as C-1/H-2, H-14, H-21; C-3/H-2, H-4, H-36; C-10/H-9, H-11, H-12, Me-18; C-13/H-12, H-14; C-23/ H-22, H-24, Me-38; etc. (Figure 1). By combination of the ¹H⁻¹H COSY and HMBC correlations (Table 1 and Figure 1), the structure of 1 could be deduced. The relative configurations of the stereogenic carbons of 1, depicted in Figure 2, were determined from 2D NOESY correlation peaks. The isopropyl group at C-12 had the same orientation as the methyl ester group at C-1. Thus the relative configuration of C-12 is R^* , as referred to the lowest carbon (C-1) having *R*-chirality. The 22*E*, 27*E*, and 34*Z* configurations of the olefinic bonds were deduced by the ¹³C NMR chemical shifts of the olefinic methyl groups. The absolute configuration of 1 remains undetermined.

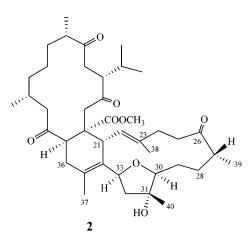
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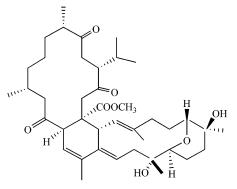
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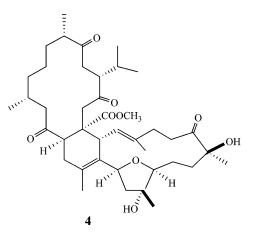




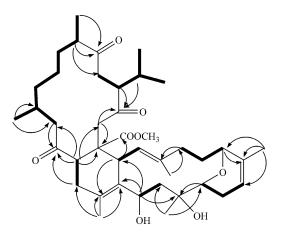








A suitable single crystal of 1 was obtained from EtOH, and X-ray diffraction analysis was used to confirm the structure and determine the relative configuration. The



¹H-¹H COSY \longrightarrow HMBC H \longrightarrow C Figure 1. ¹H-¹H COSY and key HMBC correlations of 1.

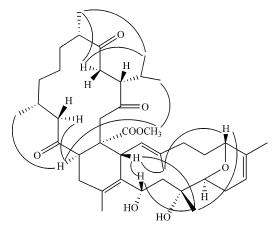


Figure 2. NOEs observed in the NOESY of 1.

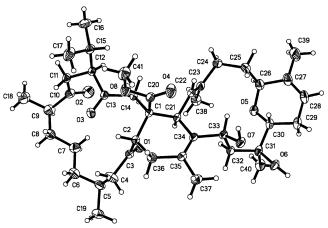


Figure 3. Perspective drawing of the X-ray structure of methyl tortuoate A (1).

results are shown in Figure 3 and in the Experimental Section. Similar to the structure of methyl sartotuate, X-ray showed that H-2 and the methyl ester have the same orientation; that is, they are cis to each other. The relative configuration of the 10 chiral carbons was determined as $1R^*$, $2R^*$, $5R^*$, $9S^*$, $12R^*$, $21R^*$, $26R^*$, $30S^*$, $31R^*$, $33R^*$.

Compound **2** was obtained as colorless crystals, mp 229–231 °C, $[\alpha]^{25}_{D}$ +84° (*c* 0.031, CH₃OH). HRFABMS [*m/z* 705.4577 (M + Na)⁺] revealed a molecular formula of C₄₁H₆₂O₈. Its UV spectrum did not show any conjugated system. The IR spectrum (KBr) indicated the presence of hydroxyl (3506, 1071 cm⁻¹), ester (1740, 1210 cm⁻¹), and carbonyl groups (1705 cm⁻¹). The NMR spectra of **2** (Table

Table 1. $^{1}\mathrm{H}$ (500 MHz) and $^{13}\mathrm{C}$ (125 MHz) NMR Data of Compounds 1 and 2 (CDCl_3)

	1			2		
C/H	$^{13}\mathrm{C}~\delta$ m	$^1\!\mathrm{H}~\delta~(\mathrm{m},J~\mathrm{in}~\mathrm{Hz})$	¹ H ⁻¹ H COSY	$^{13}\mathrm{C} \ \delta \ \mathrm{m}$	$^{1}\mathrm{H}~\delta~(\mathrm{m},J~\mathrm{in}~\mathrm{Hz})$	$^{1}\mathrm{H}{-}^{1}\mathrm{H}\mathrm{COSY}$
1	$49.2 \mathrm{~s}$			$50.5 \mathrm{~s}$		
2	46.6 d	3.82 (1H, dd, 5.0, 6.6)	H-36	43.4 d	3.51 (1H, m)	H-36
3	$212.0 \mathrm{~s}$			$208.6 \mathrm{~s}$		
4	$50.4 \mathrm{t}$	2.60 (1H, dd, 19.0, 8.0)	H-5	$53.8\ 7$	2.38 (1H,dd, 20.0, 2.0)	H-5
		2.29 (1H, dd, 19.0, 4.0)	H-5		3.16 (1H,dd, 20.0, 10.0)	H-5
5	26.4 d	2.10 (1H, m)	H-4, H-6, H-19	27.4 d	1.74 ((1H, m)	H-4, H-6, H-19
6	36.4 t	1.16 (2H, m)	H-5, H-7	$37.2~{ m t}$	1.15 (2H, m)	H-5, H-7
7	32.5 t	1.80 (2H, m)	H-6, H-8	$32.5 \mathrm{t}$	1.36 (1H, m)	H-6, H-8
					0.55 (1H, t, 12.0)	H-7, H-9
8	32.9 t	1.43 (1H, m)	H-7, H-9	$33.8~{ m t}$	1.51 (2H, m)	
		1.60 (1H, m)				
9	47.3 d	2.39 (1H, m)	H-8, H-18	47.5 d	2.48 (1H, m)	H-8, H-18
10	$213.6 \mathrm{~s}$			$213.8 \mathrm{~s}$		
11	$35.5 \mathrm{t}$	2.16 (1H, d, 16.5) 2.89 (1H, m)	H-12	$31.4 \mathrm{~s}$	1.88 (2H, dd, 17.5, 2.0)	H-12
12	50.9 d	2.94 (1H, m)	H-11, H-15	51.6 d	3.03 (1H, m)	H-11, H-15
13	$212.6 \mathrm{~s}$			$213.5 \mathrm{~s}$		
14	$47.4 \mathrm{~t}$	2.29 (1H, dd, 19.0, 8.0)		$45.8 \mathrm{t}$	2.48 (2H, d, 11.5)	
		2.60 (1H, dd, 19.0, 4.0)				
15	29.1 d	1.94 (1H, m)	H-12, H-16, H-17	28.9 d	2.22 (1H, m)	H-12, H-16, H-17
16	18.5 q	0.80 (3H, d, 7.0)	H-15	17.6 q	0.70 (3H, d, 7.0)	H-15
17	20.0 q	0.94 (3H, d, 7.0)	H-15	21.2 q	0.99 (3H, d, 7.0)	H-15
18	17.0 q	1.14 (3H, d, 7.0)	H-9	17.4 q	1.13 (3H, d, 7.0)	H-9
19	21.7 q	0.89 (3H, d, 6.5)	H-5	22.2 q	0.86 (3H, d, 7.0)	H-5
20	$174.8 \mathrm{~s}$			$174.4 \mathrm{~s}$		
21	42.1 d	3.62 (1H, d, 10.0)	H-22	45.7 d	2.50 (1H, d, 10.5)	H-22
22	122.9 d	5.12 (1H, d, 10.0)		127.4 d	4.89 (1H, d, 10.5)	
23	$141.0 \mathrm{~s}$			$136.5 \mathrm{~s}$		
24	38.3 t	1.86 (1H, m)	H-25	30.3 t	2.24 (2H, m)	H-25
		2.53 (1H, m)				
25	$24.0 \mathrm{t}$	1.02 (2H, m)	H-24, H-26	$37.7 \mathrm{t}$	2.24 (1H, m)	H-24
					2.76 (1H, m)	
26	79.7 d	4.01 (1H, br.d, 6.5)	H-25	$215.9 \mathrm{~s}$		
27	$134.4 \mathrm{s}$			49.0 d	2.37 (1H, m)	H-28, H-39
28	120.2 d	5.55 (1H, dd, 4.5, 1.5)	H-29	31.5 t	1.36 (2H, m)	H-27, H-29
29	24.7 t	2.02 (1H, m)	H-28, H-30	$25.4~{ m t}$	1.10 (2H, m)	H-28, H-30
		2.14 (1H, m)	H-28, H-30		/	
30	68.3 d	3.67 (1H, dd, 6.6, 5.0)	H-29	87.0 d	3.47 (1H, m)	H-29
31	74.2 s		** •••	80.2 s		TT 00
32	$42.2 \mathrm{~t}$	1.87 (1H, m)	H-33	$42.1~{ m t}$	1.73 (2H, m)	H-33
0.0	FO 1 1	1.60 (1H, m)	TT 00	FF 0 1		II 00
33	70.1 d	4.81 (1H, br d, 9.5)	H-32	75.3 d	5.05 (1H, dd, 10.5, 6.0)	H-32
34	132.8 s			135.2 s		
35	129.1 s		TI O	126.1 s		
36	34.6 t	2.01 (1H, m)	H-2	$33.1~{ m t}$	2.13 (2H, dd, 19.0, 9.0)	
0.7	10.0	2.54 (1H, m)		10.0	1.00 (011)	
37	19.9 q	1.91 (3H, s)		19.0 q	1.83 (3H, s)	
38	19.8 q	1.86 (3H, s)		19.2 q	1.64 (3H, s)	II 07
39	20.3 q	1.64 (3H, s)		17.4 q	1.08 (3H, d, 7.0)	H-27
40	23.1 q	1.09 (3H, s)		23.5 q	1.28 (3H, s)	
41	51.4 q	3.57 (3H, s)		$51.1~{ m q}$	3.48(3H,s)	

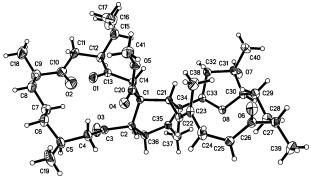


Figure 4. Perspective drawing of the X-ray structure of methyl tortuoate B (2).

1) showed one methyl ester ($\delta_{\rm C}$ 174.4, C-20; 51.1, C-41; $\delta_{\rm H}$ 3.60, 3H, s), four ketone carbonyl groups ($\delta_{\rm C}$ 215.9, C-26; 213.8, C-10; 213.5, C-13; and 208.6, C-3), one trisubstituted

and one tetrasubstituted double bond ($\delta_{\rm C}$ 126.1, C-35; 135.2, C-34; 136.5, C-23; 127.4, C-22; $\delta_{\rm H}$ 4.89, d, J = 10.5Hz, H-22). In addition, one tertiary hydroxyl ($\delta_{\rm C}$ 80.2, C-31) and an ether linkage ($\delta_{\rm C}$ 87.0, C-30; 75.3, C-33) were revealed. The ¹³C NMR DEPT, HMQC, and ¹H-¹H COSY led to the assignment of all H and C signals. A singlecrystal X-ray diffraction analysis (Figure 3 and Experimental Section) was performed, and the relative configuration of 2 was also determined. There are 10 chiral carbons in 2; the relative configuration of the stereocenters $(1R^*, 2R^*, 5R^*, 9S^*, 12R^*, 21R^*, 27R^*, 30S^*, 31R^*, 33S^*)$ was almost the same as those of 1, except for $27R^*$ and $33S^*$ in 2 instead of $26R^*$ and $33R^*$ in 1. Its structure was similar to that of compound 4, except for the absence of one tertiary hydroxyl group at C-27. It is worthy to note that the orientation of the isopropyl group relative to the ring junction in 1 and 2 is the same as for methyl sartortuate,⁵ methyl isosartortuate,⁶ and methyl neosartortuate acetate.9

Methyl tortuoate A (1) and methyl tortuoate B (2) exhibited in vitro cytotoxicity against the human nasophyringeal carcinoma CNE-2 cell line, with IC_{50} values of 22.7 and 24.7 $\mu\text{g/mL},$ and the murine P-388 tumor cell line, with $\rm IC_{50}$ values of 3.5 and 5.0 $\mu g/mL$, respectively.

The known compound 3 was identified by comparison with literature data.⁵

Experimental Section

General Experimental Procedures. The melting points were determined using an X-6 micromelting point apparatus and are uncorrected. ¹H and ¹³CNMR spectra were recorded at 500 and 125 MHz, respectively, in CDCl₃ using TMS as internal standard. Optical rotations were measured on a Schmidt+Haensch Polaptronic hngw5 polarimeter. The X-ray diffraction data were collected on a Bruker SMART 1000 CCD X-ray diffractometer. Si gel (200-300 mesh) was used for column chromatography.

Animal Material. The soft coral S. tortuosum Tix.-Dur. (Alcyoniidae) was collected from Sanya Bay, Hainan Island, China, in 2002 and was identified by Dr. Zhican Tang, Institute of Oceanology, Academia Sinica. A voucher specimen is deposited in the Research Center of Organic Natural Products, Sun Yat-sen University, Guangzhou, China.

Extraction and Isolation. The soft coral S. tortuosum (300 g, dried wt) was extracted with CH₂Cl₂-MeOH (1:1) to give a brown extract, which was extracted with petroleum ether and then with EtOAc. The residue in EtOAc was chromatographed on silica gel. Fraction A, eluted with petroleum ether-EtOAc (3:1), was further purified by chromatography on silica gel eluting with a gradient elution of EtOAc-MeOH to afford methyl tortuoate A (1) (13 mg). Fraction B, eluted with petroleum ether-EtOAc (2:1), after further purification, gave methyl tortuoate B (2) (18 mg). Fraction C, eluted with petroleum ether-EtOAc (1:1), after chromatography twice on silica gel, gave methyl sartortuoate (3) (5 mg).

Methyl tortuoate A (1): colorless prisms, mp 242-244 °C (EtOH); $[\alpha]^{25}_{D}$ +345° (c 0.025, CHCl₃); IR (KBr) ν_{max} 3381, 1742, 1707, 1194, 1104 cm⁻¹; ¹H NMR and ¹³C NMR data in Table 1; HRFABMS m/z 705.4359 (calcd for C41H62O8Na,- $705.4342 \,[M + Na]^+$).

X-ray Crystal Data for 1. Crystal data were as follows: colorless crystal, $C_{41}H_{62}O_8$, fw 682.91; crystal dimensions 0.50 $\times 0.28 \times 0.26$ mm; crystal system monoclinic; space group P2₁; unit cell dimensions a = 11.6213(15) Å, $\alpha = 90^{\circ}$, b = 9.9065-(12) Å, $\beta = 92.201(2)^\circ$, c = 16.638(2) Å, $\gamma = 90^\circ$; Z = 2; V =1914.1(4) Å³; calculated density 1.185 g/cm³; $F_{000} = 744$. The R (R_w) value of 1 was 0.0402 (0.1106). The X-ray measurements were made on a Bruker SMART 1000 CCD X-ray diffractometer with graphite-monochromated Mo Ka (λ 0.71073Å) radiation at 293(2) K. A total of 11 761 reflections were collected, 8048 of which were independent ($R_{\text{int}} = 0.0089$). The structure was solved by direct methods (SHELXTL-V5.0) and refined with full-matrix least-squares calculations. The final cycle of full-matrix least-squares refinement was based

on 8048 observed reflections $(I > 2\sigma(I))$ and 442 variable parameters. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were included but not refined.

Methyl tortuoate B (2): colorless prisms, mp 229-231 °C (EtOH); $[\alpha]^{25}_{D}$ +84° (c 0.03, CH₃OH); IR (KBr) ν_{max} 3506 (s), 1740, 1705, 1210, 1071 cm⁻¹; ¹H NMR and ¹³C NMR data in Table 1; HRFABMS m/z 705.4377 (calcd for $C_{41}H_{62}O_8Na$, $705.4342 \,[M + Na]^+$).

X-ray Crystal Data for 2. Crystal data were as follows: colorless crystal, C₄₁H₆₂O₈·C₂H₅OH, fw 728.97; crystal dimensions $0.50 \times 0.42 \times 0.18$ mm; crystal system orthorhombic; space group $P2_12_12_1$; unit cell dimensions a = 9.4139(15) Å, α $= 90^{\circ}, b = 9.7850(16) \text{ Å}, \beta = 90^{\circ}, c = 45.436(7) \text{ Å}, \gamma = 90^{\circ}; Z$ = 4; V = 4185.3(12) Å³; calculated density 1.157 g/cm³; $F_{000} =$ 1592. The $R(R_w)$ value of **2** was 0.0463 (0.1218). The X-ray measurements were made on a Bruker SMART 1000 CCD X-ray diffractometer with graphite-monochromated Mo K α (λ 0.71073 Å) radiation at 293(2) K. Data were collected up to 27.03° in θ . A total of 28 452 reflections were collected, 9159 of which were independent ($R_{\rm int} = 0.0199$). The structure was solved by direct methods (SHELXTL-V5.0) and refined with full-matrix least-squares calculations. The final cycle of fullmatrix least-squares refinement was based on 9159 observed reflections $(I \ge 2\sigma(I))$ and 469 variable parameters. The nonhydrogen atoms were refined anisotropically. Hydrogen atoms were included but not refined.

Cytotoxicity Bioassays. The tetrazolium-based colorimetric assay (MTT assay) was used for the in vitro assay of cytotoxicity to human nasophyringeal carcinoma (CNE-2) and murine P-388 tumor cell lines.

Acknowledgment. This work was supported by a grant from the National Natural Science foundation (No. 29932030) and a grant from the National High Technology Development Project (863 Project, No. 2001AA624030) of China.

Supporting Information Available: This material is available free of charge via the Internet at http://pubs.acs.org.

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NP0400151